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CERTIFICATE OF ANALYSIS

134924

/

24/09/2015

Client: Larry Cook & Associates PO Box 8146 TUMBIUMBI NSW 2261

Attention: Larry Cook

Sample log in details:

Your Reference: Larry Cook - Ardmore Park No. of samples: 10 Waters Date samples received / completed instructions received 24/09/2015

Analysis Details:

Please refer to the following pages for results, methodology summary and quality control data. Samples were analysed as received from the client. Results relate specifically to the samples as received. Results are reported on a dry weight basis for solids and on an as received basis for other matrices. Please refer to the last page of this report for any comments relating to the results.

Report Details:

Date results requested by: / Issue Date: 1/10/15 30/09/15 / Date of Preliminary Report: Not issued NATA accreditation number 2901. This document shall not be reproduced except in full. Accredited for compliance with ISO/IEC 17025. Tests not covered by NATA are denoted with *.

Results Approved By:

Jacinta/Hurst

Laboratory Manager



Ion Balance						
Our Reference:	UNITS	134924-1	134924-2	134924-3	134924-4	134924-5
Your Reference		BHAP 1	BHAP 5	BHAP6	BHAP 10	PHILSSPRING
Date Sampled		23/09/2015	23/09/2015	23/09/2015	23/09/2015	23/09/2015
Type of sample		Water	Water	Water	Water	Water
Date prepared	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Date analysed	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Calcium - Dissolved	mg/L	15	24	160	68	47
Potassium - Dissolved	mg/L	1.4	1	4.2	8.3	0.5
Sodium - Dissolved	mg/L	94	16	260	390	88
Magnesium - Dissolved	mg/L	9.0	75	78	94	81
Hydroxide Alkalinity (OH ⁻) as CaCO3	mg/L	<5	<5	<5	<5	<5
Bicarbonate Alkalinity as CaCO3	mg/L	27	230	390	340	420
Carbonate Alkalinity as CaCO3	mg/L	<5	<5	<5	<5	<5
Total Alkalinity as CaCO3	mg/L	27	230	390	340	420
Sulphate, SO4	mg/L	6	20	28	59	15
Chloride, Cl	mg/L	170	53	630	670	150
Ionic Balance	%	0.58	11	-0.92	2.4	-0.080
						1
Ion Balance						

IonBalance						
Our Reference:	UNITS	134924-6	134924-7	134924-8	134924-9	134924-10
Your Reference		SOUTHERN	BH 2	BH 4	BH 5	BH 6
		SPRING	00/00/0045	00/00/0045	00/00/0045	00/00/0045
Date Sampled		23/09/2015	23/09/2015	23/09/2015	23/09/2015	23/09/2015
Type of sample		Water	Water	Water	Water	Water
Date prepared	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Date analysed	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Calcium - Dissolved	mg/L	26	9.3	39	48	38
Potassium - Dissolved	mg/L	<0.5	0.6	0.8	0.9	0.9
Sodium - Dissolved	mg/L	55	82	35	31	62
Magnesium - Dissolved	mg/L	33	30	55	54	40
Hydroxide Alkalinity (OH ⁻) as CaCO3	mg/L	<5	<5	<5	<5	<5
Bicarbonate Alkalinity as CaCO3	mg/L	210	180	300	330	280
Carbonate Alkalinity as CaCO3	mg/L	<5	<5	<5	<5	<5
Total Alkalinity as CaCO3	mg/L	210	180	300	330	280
Sulphate, SO4	mg/L	1	8	4	3	6
Chloride, Cl	mg/L	81	80	48	51	81
Ionic Balance	%	-0.29	4.1	3.5	0.40	0.21

Metals in Waters - Acid extractable Our Reference: Your Reference Date Sampled Type of sample	UNITS	134924-1 BHAP1 23/09/2015 Water	134924-2 BHAP 5 23/09/2015 Water	134924-3 BHAP 6 23/09/2015 Water	134924-4 BHAP 10 23/09/2015 Water	134924-5 PHILSSPRING 23/09/2015 Water
Date prepared	-	25/09/2015	25/09/2015	25/09/2015	25/09/2015	25/09/2015
Date analysed	-	25/09/2015	25/09/2015	25/09/2015	25/09/2015	25/09/2015
Phosphorus - Total	mg/L	0.1	0.1	0.07	0.4	0.09
Metals in Waters - Acid extractable						
Our Reference:	UNITS	134924-6	134924-7	134924-8	134924-9	134924-10
Your Reference		SOUTHERN SPRING	BH 2	BH 4	BH 5	BH 6
Date Sampled		23/09/2015	23/09/2015	23/09/2015	23/09/2015	23/09/2015
Type of sample		Water	Water	Water	Water	Water
Date prepared	-	25/09/2015	25/09/2015	25/09/2015	25/09/2015	25/09/2015
Date analysed	-	25/09/2015	25/09/2015	25/09/2015	25/09/2015	25/09/2015
Phosphorus - Total	mg/L	0.07	1.5	0.3	0.3	0.9

Miscellaneous Inorganics						
Our Reference:	UNITS	134924-1	134924-2	134924-3	134924-4	134924-5
Your Reference		BHAP 1	BHAP5	BHAP6	BHAP10	PHILSSPRING
Date Sampled		23/09/2015	23/09/2015	23/09/2015	23/09/2015	23/09/2015
Type of sample		Water	Water	Water	Water	Water
Date prepared	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Date analysed	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
рН	pH Units	6.5	8.1	7.2	7.0	7.5
Electrical Conductivity	µS/cm	640	810	2,600	2,700	1,200
Miscellaneous Inorganics						
Our Reference:	UNITS	134924-6	134924-7	134924-8	134924-9	134924-10
Your Reference		SOUTHERN SPRING	BH 2	BH 4	BH 5	BH 6
Date Sampled		23/09/2015	23/09/2015	23/09/2015	23/09/2015	23/09/2015
Type of sample		Water	Water	Water	Water	Water
Date prepared	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
Date analysed	-	24/09/2015	24/09/2015	24/09/2015	24/09/2015	24/09/2015
рН	pH Units	7.5	6.8	7.6	7.6	7.7
Electrical Conductivity	µS/cm	630	630	740	760	760

MethodID	Methodology Summary
Metals-020 ICP- AES	Determination of various metals by ICP-AES.
Inorg-006	Alkalinity - determined titrimetrically in accordance with APHA latest edition, 2320-B.
Inorg-081	Anions - a range of Anions are determined by Ion Chromatography, in accordance with APHA latest edition, 4110-B.
Inorg-041	Gravimetric determination of the total solids content of water based on APHA latest edition 2540B.
Inorg-001	pH - Measured using pH meter and electrode in accordance with APHA latest edition, 4500-H+. Please note that the results for water analyses are indicative only, as analysis outside of the APHA storage times.
Inorg-002	Conductivity and Salinity - measured using a conductivity cell at 25oC in accordance with APHA latest edition 2510 and Rayment & Lyons.

Larry Cook - Ardmore Park

Client Reference: Larry Cook - Ardmore Park							
UNITS	PQL	METHOD	Blank	Duplicate Sm#	Duplicate results	Spike Sm#	Spike % Recovery
					Base II Duplicate II % RPD		
-			24/09/2 015	134924-1	24/09/2015 24/09/2015	LCS-W1	24/09/2015
-			24/09/2 015	134924-1	24/09/2015 24/09/2015	LCS-W1	24/09/2015
mg/L	0.5	Metals-020 ICP-AES	<0.5	134924-1	15 15 RPD:0	LCS-W1	101%
mg/L	0.5	Metals-020 ICP-AES	<0.5	134924-1	1.4 1.3 RPD:7	LCS-W1	124%
mg/L	0.5	Metals-020 ICP-AES	<0.5	134924-1	94 92 RPD:2	LCS-W1	106%
mg/L	0.5	Metals-020 ICP-AES	<0.5	134924-1	9.0 9.0 RPD:0	LCS-W1	99%
mg/L	5	Inorg-006	⊲5	134924-1	<5∥<5	[NR]	[NR]
mg/L	5	Inorg-006	⊲5	134924-1	27 25 RPD:8	[NR]	[NR]
mg/L	5	Inorg-006	45	134924-1	<5 <5	[NR]	[NR]
mg/L	5	Inorg-006	<i><</i> 5	134924-1	27 25 RPD:8	LCS-W1	105%
mg/L	1	Inorg-081	<1	134924-1	6 6 RPD:0	LCS-W1	108%
mg/L	1	Inorg-081	<1	134924-1	170 170 RPD:0	LCS-W1	103%
%		Inorg-041	[NT]	134924-1	0.58 2.0 RPD:110	[NR]	[NR]
UNITS	PQL	METHOD	Blank	Duplicate	Duplicate results	Spike Sm#	Spike % Recovery
				511#	Base II Duplicate II % RPD		Recovery
-			25/09/2 015	134924-1	25/09/2015 25/09/2015	LCS-W1	25/09/2015
-			25/09/2	134924-1	25/09/2015 25/09/2015	LCS-W1	25/09/2015
mg/L	0.05	Metals-020 ICP-AES	<0.05	134924-1	0.1 0.1 RPD:0	LCS-W1	92%
UNITS	PQL	METHOD	Blank	Duplicate Sm#	Duplicate results	Spike Sm#	Spike % Recovery
					Base II Duplicate II % RPD		
-			24/09/2 015	134924-1	24/09/2015 24/09/2015	LCS-W1	24/09/2015
-			24/09/2 015	134924-1	24/09/2015 24/09/2015	LCS-W1	24/09/2015
pHUnits		Inorg-001	[NT]	134924-1	6.5 6.5 RPD:0	LCS-W1	101%
µS/cm	1	Inorg-002	<1	134924-1	640 630 RPD:2	LCS-W1	107%
UNITS	6 1	Dup.Sm#	Base+I	Duplicate Duplicate+%RF	Spike Sm# D	Spike % Reco	overy
_		[NT]		[NT]	134924-2	24/09/201	5
		L 1		61			~
_		INTI		INTI	13/02/-2	24/00/201	5
-		[NT]		[NT]	134924-2	24/09/201	5
- mg/L mg/L		[NT] [NT] [NT]		[NT] [NT] [NT]	134924-2 134924-2 134924-2	24/09/201 82% 113%	5
	- mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L	UNITS PQL -	UNITS PQL METHOD - . . - . . mg/L 0.5 Metals-020 ICP-AES mg/L 5 Inorg-006 mg/L 5 Inorg-006 mg/L 1 Inorg-006 mg/L 1 Inorg-081 mg/L 1 Inorg-081 mg/L 1 Inorg-081 mg/L 1 Inorg-081 mg/L 1 Inorg-041 VNITS PQL METHOD - . . mg/L 0.05 Metals-020 ICP-AES UNITS PQL METHOD - . . mg/L 0.05 Metals-020 ICP-AES . . <td< td=""><td>UNITS PQL METHOD Blank - - 24/09/2 015 24/09/2 015 - - 24/09/2 015 015 mg/L 0.5 Metals-020 ICP-AES <0.5</td> mg/L 0.5 Metals-020 ICP-AES <0.5</td<>	UNITS PQL METHOD Blank - - 24/09/2 015 24/09/2 015 - - 24/09/2 015 015 mg/L 0.5 Metals-020 ICP-AES <0.5	$ \begin{array}{ c c c c c c } \label{eq:linear_relation} & \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	UNITS POL METHOD Blank Duplicate Sm# Duplicate Base II Duplicate II% RPD - 24/09/2 015 134924-1 24/09/2015 [] 24/09/2015 - 24/09/2 015 134924-1 24/09/2015 [] 24/09/2015 mg/L 0.5 Metals-020 ICP-AES <0.5	UNITS POL METHOD Blank Duplicate Sm# Duplicate results Base II Duplicate II%RPD Spike Sm# - 24/09/2 134924-1 24/09/2015 24/09/2015 LCS-W1 - 24/09/2 134924-1 24/09/2015 24/09/2015 LCS-W1 mg/L 0.5 Metals-020 ICP-AES <0.5

		Client Referenc	e: Larry Cook - Ardn	nore Park	
QUALITY CONTROL Ion Balance	UNITS	Dup. Sm#	Duplicate Base + Duplicate + %RPD	Spike Sm#	Spike % Recovery
Magnesium - Dissolved	mg/L	[NT]	[NT]	134924-2	#
Hydroxide Alkalinity (OH ⁻) as CaCO3	mg/L	[NT]	[NT]	[NR]	[NR]
Bicarbonate Alkalinity as CaCO3	mg/L	[NT]	[NT]	[NR]	[NR]
Carbonate Alkalinity as CaCO3	mg/L	[NT]	[NT]	[NR]	[NR]
Total Alkalinity as CaCO3	mg/L	[NT]	[NT]	[NR]	[NR]
Sulphate, SO4	mg/L	[NT]	[NT]	134924-2	112%
Chloride, Cl	mg/L	[NT]	[NT]	134924-2	93%
Ionic Balance	%	[NT]	[NT]	[NR]	[NR]

Report Comments:

ION_BALANCE: # Percent recovery is not possible to report due to the high concentration of the element/s in the sample/s. However an acceptable recovery was obtained for the LCS.

Asbestos ID was analysed by Approved Identifier: Asbestos ID was authorised by Approved Signatory: Not applicable for this job Not applicable for this job

INS: Insufficient sample for this test NA: Test not required <: Less than PQL: Practical Quantitation Limit RPD: Relative Percent Difference >: Greater than NT: Not tested NA: Test not required LCS: Laboratory Control Sample

Quality Control Definitions

Blank: This is the component of the analytical signal which is not derived from the sample but from reagents, glassware etc, can be determined by processing solvents and reagents in exactly the same manner as for samples. **Duplicate**: This is the complete duplicate analysis of a sample from the process batch. If possible, the sample selected should be one where the analyte concentration is easily measurable.

Matrix Spike : A portion of the sample is spiked with a known concentration of target analyte. The purpose of the matrix spike is to monitor the performance of the analytical method used and to determine whether matrix interferences exist.

LCS (Laboratory Control Sample) : This comprises either a standard reference material or a control matrix (such as a blank sand or water) fortified with analytes representative of the analyte class. It is simply a check sample.

Surrogate Spike: Surrogates are known additions to each sample, blank, matrix spike and LCS in a batch, of compounds which are similar to the analyte of interest, however are not expected to be found in real samples.

Laboratory Acceptance Criteria

Duplicate sample and matrix spike recoveries may not be reported on smaller jobs, however, were analysed at a frequency to meet or exceed NEPM requirements. All samples are tested in batches of 20. The duplicate sample RPD and matrix spike recoveries for the batch were within the laboratory acceptance criteria.

Filters, swabs, wipes, tubes and badges will not have duplicate data as the whole sample is generally extracted during sample extraction.

Spikes for Physical and Aggregate Tests are not applicable.

For VOCs in water samples, three vials are required for duplicate or spike analysis.

Duplicates: <5xPQL - any RPD is acceptable; >5xPQL - 0-50% RPD is acceptable.

Matrix Spikes, LCS and Surrogate recoveries: Generally 70-130% for inorganics/metals; 60-140% for organics (+/-50% surrogates) and 10-140% for labile SVOCs (including labile surrogates), ultra trace organics and speciated phenols is acceptable.

In circumstances where no duplicate and/or sample spike has been reported at 1 in 10 and/or 1 in 20 samples respectively, the sample volume submitted was insufficient in order to satisfy laboratory QA/QC protocols.

When samples are received where certain analytes are outside of recommended technical holding times (THTs), the analysis has proceeded. Where analytes are on the verge of breaching THTs, every effort will be made to analyse within the THT or as soon as practicable.